

## Post-decay processes in radiopharmaceutical precursors studied via nuclear spectroscopy

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Modern radiopharmaceuticals are actively used for diagnostics and therapy in nuclear medicine. The concept of a modern radiopharmaceutical allows incorporating radionuclides with similar properties to the same precursors. Radiopharmaceutical precursor and its stability play a major role in applicability of such a drug. Post-decay processes, namely Auger and conversion electron emissions, may

cause a change in local environment which can lead to release of the daughter from the radiopharmaceutical and, therefore, difficulties in its application.

The present work highlights several medically relevant isotopes –  $^{44}\text{Sc}$  and  $^{111}\text{In}$  coupled with the chelators.  $^{44}\text{Sc}$  ( $t_{1/2} = 4.04$  h) is a radiometal with favourable decay properties for positron emission tomography (PET) and  $^{111}\text{In}$  ( $t_{1/2} = 2.8$  d) is already in the clinical use for diagnostics via single photon emission computed tomography (SPECT). Moreover,  $^{111}\text{In}$  is a well-established probe nucleus in perturbed angular correlation of  $\gamma$ - $\gamma$  rays (PAC), which allows for the study of hyperfine interactions and, hence, is a unique technique when analysing post-decay processes in radiopharmaceutical precursors. Also, the study demonstrates feasibility of  $^{44}\text{mSc}/^{44}\text{gSc}$  radionuclide generator induced by post-decay processes. The radionuclide generator yield was measured by  $\gamma$ -spectroscopy and equals to  $9.8 \pm 1.0\%$ . This result indicates the influence of post-decay processes on the initial chelate complexes for the elements with medium  $Z$  and significantly changes the overall trend.

**Notes:**